# ARISTOLODIONE, A 4,5-DIOXOAPORPHINE FROM ARISTOLOCHIA CHILENSIS

## ALEJANDRO URZÚA,<sup>1</sup> ALAN J. FREYER, and MAURICE SHAMMA

### Department of Chemistry, The Pennsylvania State University, University Park, Pennsylvania 16802

The 4,5-dioxoaporphines are a small group of aporphinoids found mostly among members of the botanical families Menispermaceae. Berberidaceae, Annonaceae, Fumariaceae, and Aristolochiaceae (1). Presently, an investigation of the leaves and stems of Aristolochia chilensis Miers (Aristolochiaceae) supplied the known 4,5dioxoaporphines cepharadione-A [1] (2) and 4,5-dioxodehydroasimilobine [2] (3), and the new aristolodione [3].

Although the ir, uv, and mass spectra of our cepharadione-A were similar to those already published for this alkaloid (2,4,5), our <sup>1</sup>H-nmr spectrum, obtained at 360 MHz (CDCl<sub>3</sub>), was better defined than that previously reported (2) and has been summarized around expression 1. In particular, we were able to make specific assignments for the aromatic absorptions through the use of nmr nOe studies. Irradiation of the N-methyl singlet at  $\delta$  3.87 led to a 15.8% enhancement of the  $\delta$  7.54 singlet which must, therefore, be assigned to H-7. The other aromatic singlet absorption, which appears at  $\delta$  8.15, thus represents H-3.

The new phenolic alkaloid, aris-

tolodione [3],  $C_{18}H_{13}NO_4$ , was obtained as an optically inactive amorphous orange powder. The 360 MHz (DMSO- $d_6$ ) nmr spectrum is quoted around expression 3. An N-methyl singlet appears as  $\delta$  3.72; an 0-methyl singlet is found further downfield at  $\delta$ 4.03.

Again, as in the case of cepharadione-A, chemical shift assignments were confirmed by nOe studies. Irradiation of the N-methyl signal ( $\delta$  3.72) led to a 14.3% enhancement of H-7 ( $\delta$  7.67) so that these protons must be proximate. Additionally, irradiation of the 0-methyl signal ( $\delta$  4.03) resulted in a 10.7% enhancement of the most downfield aromatic multiplet ( $\delta$  9.46) representing H-11. It follows that the methoxyl group is located at C-1 rather than at C-2. By the same token, the phenolic function must be bonded at the C-2 site.

With the clarification of the <sup>1</sup>H-nmr spectrum of aristolodione [3], the nmr spectrum of the related known alkaloid 4,5-dioxodehydroasimilobine [2] could be readily interpreted and is given around expression 2. It will be noted that in the <sup>1</sup>H-nmr spectra of the three alkaloids studied here, the most down-



<sup>1</sup>Permanent address: Departamento de Química, Universidad de Santiago de Chile, Santiago 2, Chile.



# **EXPERIMENTAL**

PLANT COLLECTION, EXTRACTION, AND ISOLATION .- A. chilensis (160 g, dry leaves and stems) was collected in Lo Prado Pass, 6 miles from Santiago, Chile, in November 1985. Voucher specimens were deposited in the herbarium of the Natural History Museum in Santiago. The plant was dried, powdered, and extracted successively with petroleum ether and EtOH at room temperature. The EtOH extracts were concentrated and partitioned between CHCl<sub>3</sub> and 3% aqueous NaHCO3. The organic layer was stripped of solvent and fractionated by column chromatography over Si gel, using CHCl3 gradually enriched with MeOH. The fractions were monitored by tlc on Si gel, using the systems CHCl3-MeOH (95:5) and C6H6-EtOAc (3:1) and spraying with Dragendorff's reagent. Final purification of the alkaloids was on Si gel tlc glass plates using the system CHCl<sub>3</sub>-MeOH (95:5).

CEPHARADIONE-A [1].—Amorphous orange solid, 2 mg; uv  $\lambda$  max (MeOH) 234, 276, 286, 301, 313, 397, 430 nm (log  $\epsilon$  4.60, 4.39, 4.39, 4.15, 4.16, 3.96, 3.89); uv  $\nu$  max (CHCl<sub>3</sub>) 1657 cm<sup>-1</sup>; eims *m*/*z* 305 (M<sup>+</sup>, 54%), 277 (100), 260 (17), 248 (17), 219 (12), 163 (41), 150 (13), 138 (7).

### 4,5-DIOXODEHYDROASIMILOBINE [2].—

Amorphous orange solid, 2 mg. Spectrally identical with data reported in the literature (4).

ARISTOLODIONE [3].—Amorphous orange solid, 1.5 mg; uv  $\lambda$  max (MeOH) 239, 273, 290, 302, 314, 438 nm (log  $\epsilon$  4.70, 4.34, 4.18, 4.26, 4.28, 4.18);  $\lambda$  max (MeOH+OH<sup>-</sup>) 259, 314, 323, 395, 484 nm (log  $\epsilon$  4.59, 4.24, 4.23, 3.94, 4.01); eims *m*/*z* 307 (M<sup>+</sup>, 100%), 279 (62), 264 (97), 236 (50).

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